

## NanoParticle

<b>Title</b>	<b>Magnetic nanoparticles: Essential factors for sustainable environmental applications</b>
<b>Author Name</b>	Samuel C.N. Tang, Irene M.C. Lo
<b>Journal Name</b>	Water Research
<b>Year</b>	2013
<b>Volume and Issue</b>	Volume 47, Issue 8,
<b>Pages</b>	2613–2632
<b>Abstracts</b>	In recent years, there has been an increasing use of engineered magnetic nanoparticles for remediation and water treatments, leading to elevated public concerns. To this end, it is necessary to enhance the understanding of how these magnetic nanoparticles react with contaminants and interact with the surrounding environment during applications. This review aims to provide a holistic overview of current knowledge of magnetic nanoparticles in environmental applications, emphasizing studies of zero-valent iron (nZVI), magnetite (Fe <sub>3</sub> O <sub>4</sub> ) and maghemite (γ-Fe <sub>2</sub> O <sub>3</sub> ) nanoparticles. Contaminant removal mechanisms by magnetic nanoparticles are presented, along with factors affecting the ability of contaminant desorption. Factors influencing the recovery of magnetic nanoparticles are outlined, describing the challenges of magnetic particle collection. The aggregation of magnetic nanoparticles is described, and methods for enhancing stability are summarized. Moreover, the toxicological effects owing to magnetic nanoparticles are discussed. It is possible that magnetic nanoparticles can be applied sustainably after detailed consideration of these discussed factors.
<b>Keywords</b>	

<b>Title</b>	<b>Silica nanoparticles capture atmospheric lead: Implications in the treatment of environmental heavy metal pollution</b>
<b>Author Name</b>	Xifei Yanga, Zhiguo Shenb, Bing Zhanga, Jianping Yangc, Wen-Xu Honga, Zhixiong Zhuanga, Jianjun Liu
<b>Journal Name</b>	Chemosphere
<b>Year</b>	2013
<b>Volume and Issue</b>	90
<b>Pages</b>	653–656
<b>Abstracts</b>	Lead (Pb) contamination in the air is a severe global problem, most notably in China. Removal of Pb from polluted air remains a significant challenge. It is unclear what potential effects silica nanoparticles (SiNPs) exposure can have on atmospheric Pb. Here we first characterized the features of SiNPs by measuring the particle size, zeta potential and the specific surface area of SiO <sub>2</sub> particles using a Nicomp 380/ZLS submicron particle sizer, the Brunauer – Emmett–Teller (BET) method and transmission electronic microscopy (TEM). We measured the content of the metal Pb adsorbed by SiNPs exposed to two Pb polluted electric battery plants using inductively coupled plasma mass spectrometry (ICP-MS). It is found that SiNPs exposed to two Pb polluted electric battery plants absorb more

	atmospheric Pb compared to either blank control or micro-sized SiO <sub>2</sub> particles in a time-dependent manner. This is the first study demonstrating that SiNPs exposure can absorb atmospheric Pb in the polluted environment. These novel findings indicate that SiNPs have potential to serve as a significant adsorbent of Pb from industrial pollution, implicating a potentially novel application of SiNPs in the treatment of environmental heavy metal pollution.
<b>Keywords</b>	

<b>Title</b>	<b>Investigation of gold nanoparticles uptake and their tissue level distribution in rice plants by laser ablation-inductively coupled-mass spectrometry</b>
<b>Author Name</b>	Jeremy Koelmel, Thomas Leland, Huanhua Wang, Dulasiri Amarasiriwardena, Baoshan Xing
<b>Journal Name</b>	Environmental Pollution
<b>Year</b>	2013
<b>Volume and Issue</b>	174
<b>Pages</b>	222-228
<b>Abstracts</b>	Lead (Pb) contamination in the air is a severe global problem, most notably in China. Removal of Pb from polluted air remains a significant challenge. It is unclear what potential effects silica nanoparticles (SiNPs) exposure can have on atmospheric Pb. Here we first characterized the features of SiNPs by measuring the particle size, zeta potential and the specific surface area of SiO <sub>2</sub> particles using a Nicomp 380/ZLS submicron particle sizer, the Brunauer–Emmett–Teller (BET) method and transmission electronic microscopy (TEM). We measured the content of the metal Pb adsorbed by SiNPs exposed to two Pb polluted electric battery plants using inductively coupled plasma mass spectrometry (ICP-MS). It is found that SiNPs exposed to two Pb polluted electric battery plants absorb more atmospheric Pb compared to either blank control or micro-sized SiO <sub>2</sub> particles in a time-dependent manner. This is the first study demonstrating that SiNPs exposure can absorb atmospheric Pb in the polluted environment. These novel findings indicate that SiNPs have potential to serve as a significant adsorbent of Pb from industrial pollution, implicating a potentially novel application of SiNPs in the treatment of environmental heavy metal pollution.
<b>Keywords</b>	

<b>Title</b>	<b>The greener synthesis of nanoparticles</b>
<b>Author Name</b>	Oxana V. Kharissova, H.V. Rasika Dias, Boris I. Kharisov, Betsabee Olvera, Pe´rez, and Victor M. Jimenez Pe´rez
<b>Journal Name</b>	Trends in Biotechnology
<b>Year</b>	2013
<b>Volume and Issue</b>	31, No. 4
<b>Pages</b>	
<b>Abstracts</b>	In this review, we examine ‘greener’ routes to nanoparticles of zerovalent metals, metal oxides, and salts with an emphasis on recent developments. Products from nature or those

	<p>derived from natural products, such as extracts of various plants or parts of plants, tea, coffee, banana, simple amino acids, as well as wine, table sugar and glucose, have been used as reductants and as capping agents during synthesis. Polyphenols found in plant material often play a key role in these processes. The techniques involved are simple, environmentally friendly, and generally one-pot processes. Tea extracts with high polyphenol content act as both chelating/reducing and capping agents for nanoparticles. We discuss the key materials used in the field: silver, gold, iron, metal alloys, oxides, and salts.</p>
<b>Keywords</b>	

<b>Title</b>	<b>Synthesis of metallic nanoparticles using plant extracts</b>
<b>Author Name</b>	Amit Kumar Mittal, Yusuf Chisti, Uttam Chand Banerjee
<b>Journal Name</b>	Biotechnology Advances
<b>Year</b>	2013
<b>Volume and Issue</b>	Volume 31, Issue 2,
<b>Pages</b>	346–356
<b>Abstracts</b>	<p>Biomolecules present in plant extracts can be used to reduce metal ions to nanoparticles in a single-step green synthesis process. This biogenic reduction of metal ion to base metal is quite rapid, readily conducted at room temperature and pressure, and easily scaled up. Synthesis mediated by plant extracts is environmentally benign. The reducing agents involved include the various water soluble plant metabolites (e.g. alkaloids, phenolic compounds, terpenoids) and co-enzymes. Silver (Ag) and gold (Au) nanoparticles have been the particular focus of plant-based syntheses. Extracts of a diverse range of plant species have been successfully used in making nanoparticles. In addition to plant extracts, live plants can be used for the synthesis. Here we review the methods of making nanoparticles using plant extracts. Methods of particle characterization are reviewed and potential applications of the particles in medicine are discussed.</p>
<b>Keywords</b>	